

V.A.

24902 KISELEV, V. A. Obshcheye Uravmeniye Verevochoy Krivoy Pri Deystvii Vertikal noy Nagruzki. Trudy Mosk. Avtomob-dor. In-ta Im. Molotova, SO: Letopis', No.33, 1949

KISELEV, V. A.

APPRING MODELLEY, V. A. O Forme Raynovesiya Tyazhelpx kidp86c00513R000722810011-9"
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SO: Letopis', No.33, 1949

KISELËV, V. A.

TREASURE ISLAND BIBLIOGRAPHICAL REPORT

AID 535 - I

PHASE I

Call No.: AF603914

Author: KISELEV, V. A., Prof., Dotsent of Tech. Sci.

Full Title: DYNAMIC INFLUENCE LINES OF THE BENDING MOMENT AND OF THE BOOK

K voprosu o dinamicheskikh linyakh vliyaniya 1zgibayushchego momenta i poperechnoy sily Transliterated Title:

PUBLISHING DATA

Originating Agency: Moscow Institute of Railroad Transport Engineers
Originating Agency: Moscow Institute of Railroad Transport Engineers
im. Stalin (MIIT), Trudy, Issue 76, Construction Mechanics
Publishing House: State Publishing House of Railroad Transport
Publishing House: No. 25 (108-122)

Date: 1952

Editor-in-Chief: Litvin, G. A., Kand. of Tech. Sci. Editors: Profs., Doc. of Tech. Sci. Prokof yev, I. P., Editorial Staff

Pratusevich, Ya. A., and Sinel'nikov, V. V.
Others: The preface was written by Gerasimov, A. S., Chief of MIIT,

PURPOSE: A paper intended for engineering-technical and

scientific Workers of railroad transport.

Coverage: On the basis of the work of the academician Krylov, A. N., TEXT DATA

CIA-RDP86-00513R000722810011-**APPROVED FOR RELEASE: 09/17/2001** K voprosu o dinamicheskikh linyakh vliyaniya izgibayu-

shchego momenta i poperechnoy sily v balkakh

the author gives a method for the determination of the influence line of the bending moment and of the lateral force in a beam, due to a mass-less load moving with a determined speed. The analyzis of some basic dynamic coefficients are given. The author divides his article as follows: 1. Introduction; 2. Solution of acad. A. N. Krylov; 3. Dynamic coefficients; 4. Dynamic influence lines; 5. Auxiliary tables; 6. Movement of the band load; 7. Influence lines due to load periodically changing in time. Formulae, tables

No. of References: Total - 5, Russian 4, dated 1905-1939. Other 1,

Facilities: Names of several scientists working in the field of determination of stresses due to moving loads are mentioned in the text.

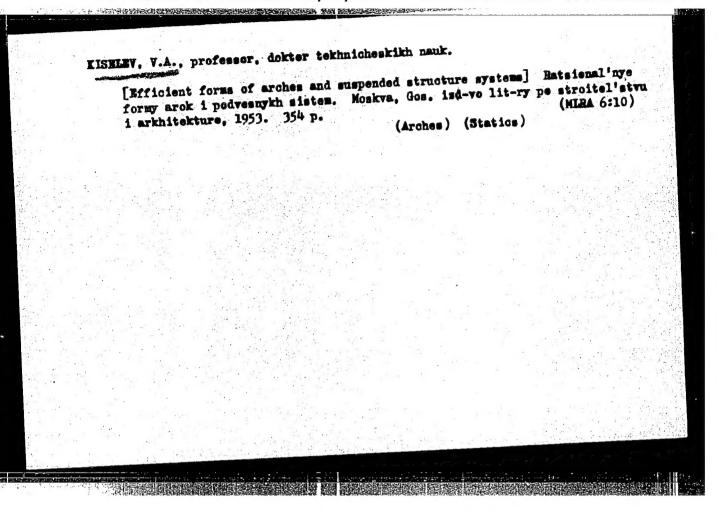
KISELEV, V. A.

RUBIEIN, M.V.: KISKLEY, V.A., doktor tekhnicheskikh nauk, professor, retsengent; PUPU, A.A., doktor tekhnicheskikh nauk, professor, retsengent; LIVIT, M.A., dotsent, redaktor.

[Manual for practical studies on the strength of materials]
Rukovodstvo k prakticheskim saniatiiam po sepretivleniiu materialov. isd. 2-e. ispr. i dep. Moskva, Ges. nauchno-tekhn. isd-ve malniostroitol nai lit-ry. 1953. 307 p.

(Strength of materials)

(Strength of materials)



ence and inventions as	Prizes (of the Council of Mini mounces that the following so as have been submitted for con (Sovetaknya Kultura, Moscow,	metition for Stalin Prizes for No. 22-40, 20 Feb - 3 Apr 1954)
New	Title of Work	Rominated by
Kiselev, V.A.	"Rational Forms of Arch and Suspension Systems"	es Moscow Automobile Highway

RISELEV, V. A.

"The Rational Axis of Three-Ball Arches of Underwater Tunnels of Constant Cross Section," Dokl. AN SSSR, 90, No.1, pp 45-48, 1953

Rational axis is the name given to that axis for which the bending moments on all cross sections of an arch equal zero. Derives the differential eqs for the desired rational axis of an arch, which eqs are too complicated for integration; hence gives a modification for earier solution. Presented by Acad. A.I.Nekrasov 16 Mar 53

259T100

RABINOVICH, Isaak Moiseyevich, doktor tekhnicheskikh nauk, professor;
BEULHOV, N.I., professor, doktor tekhnicheskikh nauk, retsensent;
KISHLEV, V.A., professor, doktor tekhnicheskikh nauk, retsensent.

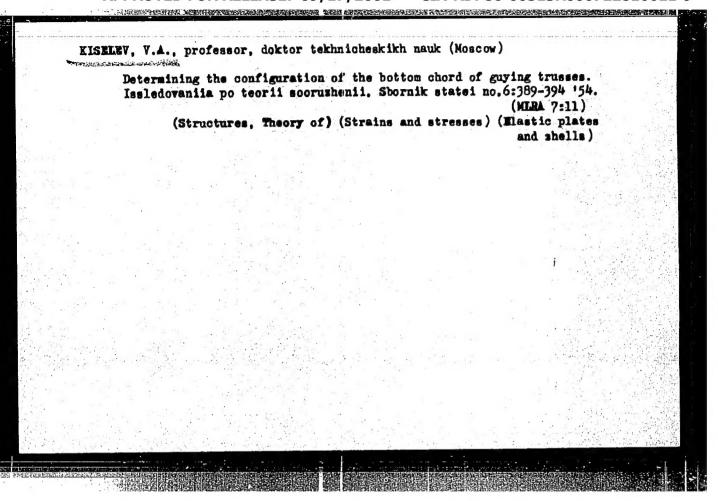
SHITKO, I.A., kandidat tekhnicheskikh nauk, nanchnyy redaktor;
TUMARKIN, D.M., redaktor; SMOL'YAKOVA, M.V., tekhnicheskiy redaktor.

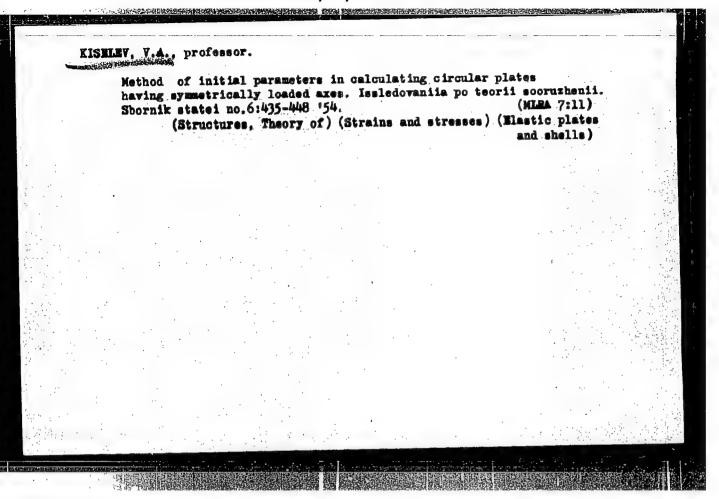
[Course in the structural mechanics of bar systems] Kurs stroitelnoi mekhaniki sterzhnavykh sistem. Part 2. [Statically indeterminate systems] Staticheski meopredelimye sistemy. Isd. 2-e, perer.
Moskva, Gos. isd-vo lit-ry po stroitel'stvu i arkhitekture. 1954,
543 p.

1. Chlen-korrespondent Akademii Hauk SSSR (for Rabinovich)

(Structures, Theory of)

CIA-RDP86-00513R000722810011-9 Kischer, VH. L'2071. Kiseler, Y. L. A method of Initial parameters for calculoting circular platter with an exially symmetrical load (in 26 Russian), isolod, fac. tearif sucsusbenti ac. 6, 135-448, 1954; Rev. no. 380, Ref. 1h. Meth 1986. The calculation of circular plates with an acially symmetrical load according to the method of mittal parameters. First of all a concentrated strip load, applied in a ring, is examined. Fuether, a case is aken of an annular load determined according to the parabolic law. Six examples are given of the calculation of complete circular places, and also with a central hole, for various particular cases of loading, including concentrated amment loads. It is shown that this solution for a circular plate is based on the case of a concentrated strip load. Using this basic solution, it is possible to obtain solutions for other forms of load as well. Generalised equations for a circular and different and more simple metrod, were obtained by S. N. Sokolov in 1935 [Tr. Mask in-et kbim masbinostt. 1935, no. 1, 64-106]. Courtesy of Referentionys Zhumal S. S. Kryukovskii, USSR Translation, courtery of Ministry of Supply, England





ITSKOVICH, G.M.; KISELEV. V.A.; CHERNAVSKIY, S.A.; BOKOV, K.H.; FAGEL',

A.Z., BONGH-OSNOLOVSKIY, M.A.; GRINCHAR, G.M.; CHERNAVSKIY, S.A.,
kandidat tekhnichenkikh nauk, nauchnyy redaktor; TIKHCHOV, A.Ya.,
tekhnicheskiy redaktor

[Collection of problems and methods of celculating machine parts]
Sbornik zadach i grimerov rascheta detalei meshin. Moskva, Gos.
nauchno-tekhn. izd-vo mashinostroit. lit-ry, 1957. 267 p. (MIRA 10:4)

(Machinery-Design)

BOKOV, Kirill Nikolayevich; ITSKOVICH, Georgiy Mikhaylovich; KISTEN, Yyacheplav Aleksandrovich; CHERNAVSKIY, Sergey Aleksandrovich; TIKHOROV, A.Ya., tekhn.red.; SOKOLOVA, T.F., tekhn.red.

[A course in designing machine parts, a textbook] Kursovoe proektirovanie detalei mashin; uchehon-spravochnoe posobie. Ind. 2-oe, perer. Moskva, Gos.nauchno-tekhn.izd-vo mashinostroit. lit-ry, 1957. 503 p. (Machinery-Design)

(Machinery-Design)

KISELEY, V.A.

AUTHORS:

Bokov, K.N., Itskovich, G.M., Kiselev, V.A.,

Chernavskiy, S.A.

TITLE:

Undergraduate Course in Design of Machine Elements. (Kursovoye proyektirovaniye detaley mashin) (Ushebno-

-sprayochnoye posobiye)

PUB. DATA:

Gosudarstvennoye nauchnortekhnicheskoye izdatel'stve

mashinostroitel noy literatury, Moscow, 1957,

2d ed. rev., 503 pp., 25,000 copies

ORIG. AGENCY: None given

EDITORS:

Ed of Publishing House: Krylov, V.I., Engr.; Science

Ed.: Itskovich, G.M., Engr.; Tech. Editors: Tikhanov, A.Ya., and Sokolova, T.F.; Corrector:

Matisen, V.G.

PURPOSE:

This book is approved by the Administration of

Special Secondary Educational Institutions, Ministry of Higher Education of the USSR, as a text for technical

Card 1/10 schools.

APPROVED FOR RELEASE: 09/17/2001 CIA-RDP86-00513R000722810011-Undergraduate Course in Design of Machine Elements. (Cont.)

COVERAGE:

The book is stated to contain the basic data and instructions for designing the drive mechanisms which are the standard subjects of study in courses in machine design at USSR technical schools. Typical design problems and calculations are given. The authors stress the importance of conducting student examinations in basically the same way as that in which students defending these are examined. Chapter XIV was written with the assistance of Bonch—Osmolovskiy, M.A., Candidate of Technical Sciences,—and Grinchar, G.N., Candidate of Technical Sciences. There are 34 references, all USSR.

Card 2/10

## CIA-RDP86-00513R000722810011-9

SOV/124-58-1-1252

Translation from: Referativnyy zhurnal, Mekhanika, 1958, Nr 1, p 156 (USSR)

Kiselev, V. A. AUTHOR:

TITLE:

On the Selection of the Cross Sections of a Flexible Thread Loaded Uniformly Along Its Length With Due Consideration of the Weight of the Thread Itself (O podbore secheniy gibkoy niti pri ravnomernoy nagruzke po yeye dline s uchetom sobstvennogo vesa niti)

PERIODICAL: V sb.: Issledovaniya po teorii sooruzheniy. Nr 7, Moscow, Gosstroyizdat, 1957, pp 597-603

Expressions are adduced for the tension, length, elongation, and necessary cross section of a thread; the expressions are based on an ABSTRACT: exact solution for the thread as a catenary line. An example is provided. I. K. Snitko

Card 1/1

CIA-RDP86-00513R000722810011-9

SOV/124-58-7-8097

Translation from: Referativnyy zhurnal, Mekhanika, 1958, Nr 7, p 113 (USSR)

AUTHOR:

Kiselaume Vr Annance

TITLE:

The Influence Coefficients in the Calculation of Continuous Beams Through Fixed-point Ratios (Koeffitsiyenty vliyaniya pri raschete nerazreznykh balok cherez fokusnyye otnosheniya)

PERIODICAL:

Tr. Mosk. avtomob.-dor. in-ta, 1957, Nr 20, pp 113-116

ABSTRACT:

Bibliographic entry

1. Beams--Mathematical analysis

Card 1/1

CHICACHTHAIR THE THE CHICAGO CONTRACTOR STREET, STREET, STREET, STREET, STREET, STREET, STREET, STREET, STREET,

CIA-RDP86-00513R000722810011-9

SOV/124-58-7-8095

Translation from: Referativnyy zhurnal, Mekhanika, 1958, Nr 7, p 113 (USSR)

AUTHOR:

Kiselev, V.A.

TITLE:

On the Differential Relationships Existing Between the Influence Lines of the Bending Moment M, the Transverse Force Q, and the Longitudinal Force N (O differentsial nykh zavisimostyakh mezhdu liniyami vliyaniya izgibayushchego momenta M, poperechnoy sily Q, i prodol noy sily N)

PERIODICAL: Tr. Mosk. avtomob.-dor. in-ta, 1957, Nr 20, pp 117-121

ABSTRACT:

Bibliographic entry

2. Materials--Stresses 1. Materials--Moments

Card 1/1

CIA-RDP86-00513R000722810011-9" **APPROVED FOR RELEASE: 09/17/2001** 

#### CIA-RDP86-00513R000722810011-9

SOV/124-58-8-9209

Translation from: Referativnyy zhurnal, Mekhanika, 1958, Nr 8, p 126 (USSR)

AUTHOR:

Kiselev, V.A.

TITLE:

Concerning Special Solutions for Some Types of Differential Equations Encountered in Structural Mechanics (O chastnykh resheniyakh nekotorykh vidov differentsial'nykh uravneniy v

stroitel noy mekhanike)

PERIODICAL:

Tr. Mosk. avtomob.-dor. in-ta, 1957, Nr 20, pp 123-126

ABSTRACT:

Bibliographic entry

Card 1/1

ITSKOVICH, G.M.; KISBIEY, Y.A.; CHERNAVSKIY, S.A., kand.tekhm.nauk;
BOKOV, K.M.; FAGEL', A.Z.; BONCH-OSMOLOVSKIY, M.A.; GRINCHAR,
G.M.; E.KHED, V.D., tekhn.red.

[Collected problems and exercises of design for the course on
machine parts] Shornik sadach i primerov reachets po kursu
detalei mashin. Isd.2-e., perer. Moskva, Gos.nauchno-tekhn.
izd-vo mashinostroit.lit-y, 1959. 330 p. (MIRA 13:10)
(Mechanical engineering--Froblems, exercises, etc.)

DYKHOVICHNYY, Abram Ionovich; RABINOVICH, I.M., prof., retsenzent; KISHLEY, V.A., prof., retsenzent; SNITKO, I.K., prof., otv.red.; PETRAKOVA, Te.P., red.izd-va; KOROVENKOVA, Z.A., tekhn.red.

[Structural mechanics: abridged course] Stroitel nais mekhanika; sokrashchennyy kurs. Izd.3.,perer. Moskva, Ugletekhizdat, 1959. 342 p. (MIRA 12:4)

1. Rukovoditel' kafedry stroitel'noy mekhaniki Voyenno-inzhenernoy akademii imeni V.V.Kuybysheva (for Rabinovich).

(Structures, Theory of)

KISELEY, V.A.

# PHASE I BOOK EXPLOITATION SOV/3453

- Chernavskiy, Sergey Aleksandrovich, Georgiy Mikhaylovich Itskovich, Vyacheslav Aleksandrovich Kiselev, Kirill Nikolayevich Bokov, Mikhail Aleksandrovich Bonch-Osmolovskiy, and Boris Pavlovich Kozintsov
- Proyektirovaniye mekhanicheskikh peredach; uchebno-spravochnoye posobiye po kursovomu proyektirovaniyu detaley mashin (Designing of Mechanical Drives; Text and Handbook On Machine Parts Designing) Moscow, Mashgiz, 1959. 740 p. 80,000 copies printed.
- Scientific Ed.: S.A. Chernavskiy; Ed. of Publishing House: N.Yu. Blagosklonova, Engineer; Tech. Ed.: A.Ya. Tikhanov; Managing Ed. for Information Literature; I.M. Monastyrskiy, Engineer.
- PURPOSE: This manual is intended for students in higher engineering schools.
- COVERAGE: This book describes the basic principles of the kinematic design of drives with a consideration of economy Card 1/8

sov/3453 Designing of Mechanical (Cont.) Fundamentals of designing speed reducers, variable speed drives, and various types of mechanical transmission are explained. Methods of designing for strength are also discussed. Examples of design and construction of drives are presented. No personalities are mentioned. There are 67 Soviet references. TABLE OF CONTENTS: 3 Poreword Assignment for a Term Project on Machine Parts (K.N. Bokov, Ch. I. 558 Engineer) The scope and content of the assignments 2. Examples of assignments Ch. II. Making Drawings and Calculation Notes Basic requirements for preparation of drawings (K.N. Bokov) Preparation and the form of calculation notes (G.M. Itsko-39 vich, Engineer) Card 2/8

BOKOV, Kirill Nikoleyevich; ITSKOVICH, Georgiy Mikhaylovich; inzh.; KISELEV,

Vyacheslav Aleksandrovich; CHERNAVSKIY, Sergey Aleksandrovich;

GIL'DENBERG, M.I., red.izd-va; MODEL', B.I., tekhn.red.

[Course in the design of machine parts; text and reference book]
Kursovoe proektirovanie detalei mashin; uchebno-spravochnoe posobie.

Izd.3. Leningrad, Gos.nauchno-tekhn.izd-vo mashinostroit.lit-ry.

1960. 507 p. (MIRA 13:11)

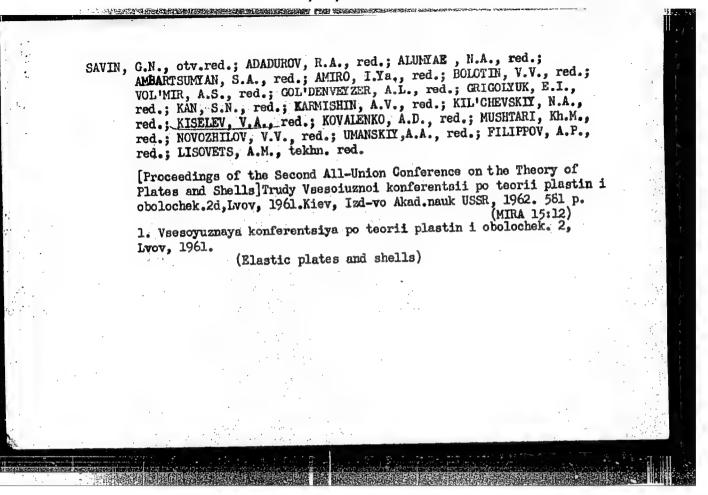
KISELEV, Vasiliy Aleksandrovich, prof., doktor tekhn.nauk; DARKOV, A.V., prof., doktor tekhn.nauk, retsenzent; FROKOF'YEV, K.N., nauchnyy red.; BORODINA, I.S., red.isd-ve; OSKUKO, L.M., tekhn.red.

[Structural mechanics] Stroitel'naia mekhanika. Moskva, Gos. izd-vo lit-ry po stroit., arkhit. i stroit. materialam, 1960.

[Structures, Theory of)

(MIRA 13:12)

Designing continuous beams for a live lead according to the limiting state, taking into account the plactic properties of the material. Issl. po teor. soorush. no.10:134-155 '61. (MIRA 14:8) (Beams and girders, Continuous)



S/879/62/000/000/044/088 D234/D308

Kiselev, V. A. (Moscow) AUTHOR:

Dynamical surfaces of influence of displacement and internal forces of orthotropic plates placed on an elastic TITLE:

base with two characteristics from a moving load having

a uniform velocity

Teoriya plastin i obolochek; trudy II Vsesoyuznoy konfe-rentsii, L'voy; 15-21 sentyabrya 1961 g. Kiev, Izd-vo AN USSR, 1962, 274-279 SOURCE:

TEXT: The author solves the problem of bending of an orthotropic

plate subject to any load varying with time

 $\mathbf{w}_{\mathbf{n}}(\mathbf{x},\mathbf{y})e^{-\alpha t}(\mathbf{a}_{\mathbf{n}}\sin \tilde{\omega}_{\mathbf{n}}t + \mathbf{b}_{\mathbf{n}}\cos \tilde{\omega}_{\mathbf{n}}t) +$ 

Card 1/2

S/879/62/000/000/044/088 D234/D308

Dynamical surfaces of ...

$$+\frac{w_{n}(x,y)}{\bar{\omega}_{n}}\int_{0}^{t}e^{-\alpha(t-u)}q_{n}(u)\sin\bar{\omega}_{n}(t-u)du$$
(16)

and derives general expressions for the surfaces of influence of displacements, also for a plate with additionally hinged edges. An expression for the critical velocity is also derived. The author includes a summary of his investigation of principal vibrations of a plate with two opposite sides hinged; these are not published because of lack of space. There is 1 figure.

Card 2/2

S/124/63/000/003/051/065 D234/D308

AUTHOR:

Kisselev, V. A.

TITLE:

PERIODICAL:

Bending of a beam beyond the yield limit, taking into account the variation of the cross-sectional dimensions

STOTI

Referativnyy zhurnal, Mekhanika, no. 3, 1963, 31, abstract 3V211 (In collection: Issled. po teorii sooruzh.

no. 11, M., Gosstroyizdat, 1962, 227-252)

TEXT: The variations of cross-sectional dimensions in both directions are taken into account. Hypotheses of plane sections and constant volume are adopted. The variation of the cross-section in the elastic stage is ignored. The author uses the notions of true (logarithmic) deformation e and true stress a (calculated from the area F of the deformed section) in the axial tension (compression). He introduces the quantity  $\psi = 1 - F/F_0$  (F<sub>0</sub> is the initial area of

the cross-section) connected with the conventional deformation &

Card 1/2

Bending of a beam ...

S/124/63/000/003/051/065 D234/D308

and with e by the relation

$$\psi = \frac{\mathcal{E}}{1 + \mathcal{E}} = \frac{\exp e - 1}{\exp e}$$

Some methods of approximating the relations s(e), s(Y), s(E) are proposed. The s(e) graph of tension (compression) is used to investigate the case of single-axis compression. A dependence between true deformations and end fibers is established and the bending moment corresponding to these deformations is calculated. The particular case of a rectangular cross-section is considered. Similar investigations are carried out using the other two relations. /Abstracter's note: Complete translation. 7

Card 2/2

KISELEY, V.A.; AFANAS'YEV, A.M., nauchn. red.; TITOVA, V.A., red.;
BARANOV, Yu.V., tekhn. red.

[Theory of external and internal forces in a bar] Teorita
vneshnikh i vnutrennikh sil brusa. IAroslav', Rosvuzizdat,
1963. 66 p.

(Beams and girders)

BAGREYEV, Vladimir Vladimirovich; VINOKUROV, Anatoliy Ivanovich;
KISELEV, Vvachealav Aleksendrovich; PANICR, Boris
Bentsionovich; ITSKOVICH, Georgi, Itahaylovich;
KONDRASHOV, D.A., insh., retsensent; RUBASHKIN, A.G.,
insh., retsensent; ARKUSHA, A.I., nauchn. red.; KOZINTSOV,
insh., retsensent; ARKUSHA, N.N., red.; YEROMITSKAYA,
B.S., nauchn. red.; VASIL'IVAV, N.N., red.; YEROMITSKAYA,
Ye.Ye., red.; SHAURAK, Ye.N., red.; KRYAKOVA, D.M., tekhn.,
red.

[Collection of problems in technical mechanics] Sbornik sadach po tekhnicheskoi mekhanike [By] V.V.Bagreev i dr. Leningrad, Sudpromgis, 1963. 551 p. (MIRA 16:8)

(Mechanical engineering—Problems, exercises, etc.)

Dynamic surfaces of effect of the displacements and internal forces of orthotropic plates on an elastic foundation with two coefficients of the bad. Issl. po teor. scorush. no. 12:43-64.

(Slastic plates and shells)

(Elastic plates and shells)

CHE-NAVSKIY, S.A., kand. tekhn.nauk; ITSKOVICH, C.M.; KISELEV, V.A.:

BOKOV, K.N.; BONCH-OSMOLOVSKIY, M.A.; KOZINTSOV, V.F.;

FEDOTOV, G.I., prof., retsenzent; GIL'DHERG, M.I., red.izdva; SOKOLOVA, T.F., tekhn. red.

[Design of mechanical transmissions] Proektirovanie mekhanicheskikh peredach; uchebno-spravochnoe posobie po kursovomu
proektirovaniiu mekhanicheskikh peredach. Izd.2., perer.
[By] S.A.Chernavskii i dr. Moskva, Mashgiz, 1963. 799 p.

(MIRA 16:12)

(Power transmissions)

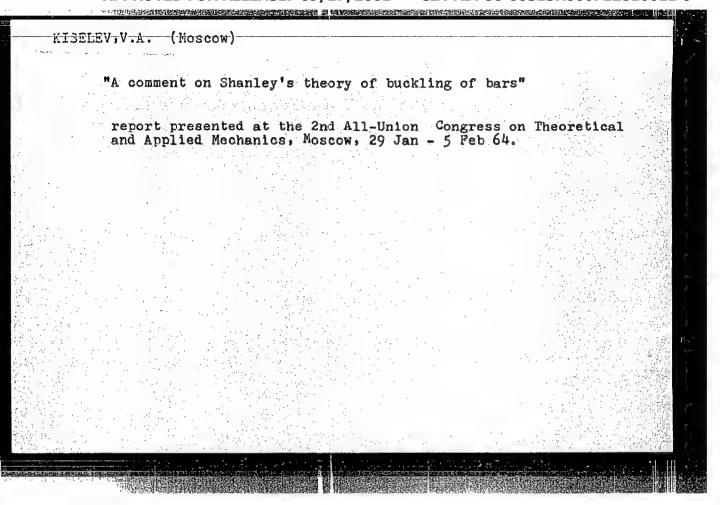
AFANAS'YEV, A.M.; YERMOLENKO, V.A.; KISELEV, V.A., zasl. deystel'
nauki i tekhniki RSFSR, dartor tekhn. nauk, prof.;
MEDHIKOV, 1.A.; OVSTANNIKOVA, M.V.; SLODODCHIKOV, A.Ya.;
TYAZHELOV, N.N.; FEDOROV, Yu.P.; TSVEY, I.Yu.; DARKOV,
A.V., doktor tekhn.nauk, prof., retsenzent; FEDOROV, Yu.P.,
kand. tekhn. nauk, nauchn. red.

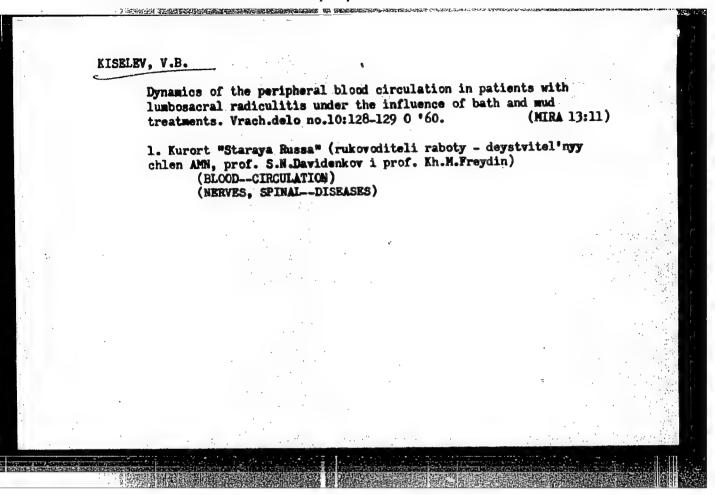
[Structural mechanics in examples and problems] Stroitel'naia mekhanika v primerakh i zadachakh. Moskva, Stroinizdat, 1964. 341 p.

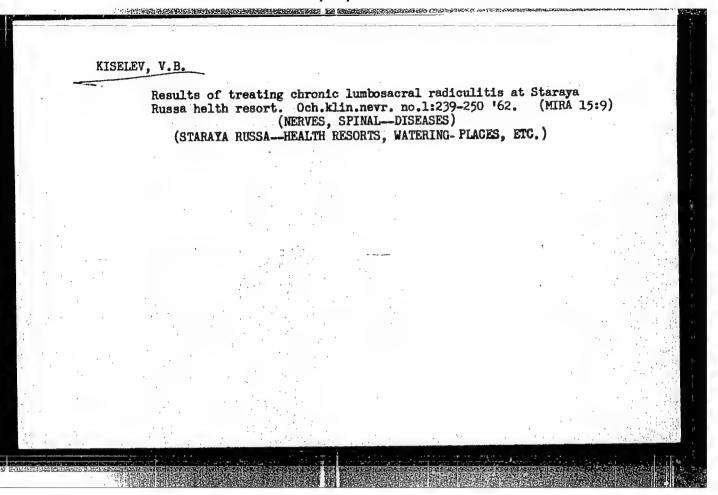
(NIRA 18:1)

KISELEV, Vasiliy Aleksandrovich, doktor tekhn. nauk, prof.;
REKACH, V.G., doktor tekhn.nauk, retsenzent;
MEDNIKOV, I.A., kand. tekhn. nauk, dots., nauchm.red.

[Structural mechanics; a special course (the dynamics and rigidity of structures)] Stroitel'naia mekhanika; spetsial'nyi kurs (dinamika i ustoichivost' socruzhenii). Moskva, Stroitzdat, 1964. 331 p. (MIRA 18:2)







MISSLEV, V. F.

"Investigation of Deformations in the Process of Cutting Cast Iron." Thesis for degree of Cand. Technical Sci. Sub 22 Feb 50, Moscow Machine Tool (and Tool) Inst imeni I. V. Stalin

Summary 71, 4 Sep 52, Dissertations Presented for Degrees in Science and Engineering in Moscow in 1950. From Vechernyaya Moskva, Jan-Dec 1950.

DEMICHEY, A.D.; KISELEY, V.F. starshiy dorozhnyy master (stantsiya Ira-Iol'
Pechorskoy dorogi); COLINGETT, A.D.; COMMUNIS, A.A.; starshiy dorozhnyy master
(Stantsiya Polotsk Belorusskoy dorogi); RUES, V.G., brigadir puti(stantstya Cheremkhov Vostochno-Sibirskoy dorogi); PAVLOV, V.M., brigadir
puti (stantsiya Cheremkhovo Vostochno-Sibirskoy dorogi); SHAKHBALAYEV,
A.M., dorozhnyy master (stantsiya Zenseli Ordzhonikidzevekoy dorogi);
TARASENKO, V.Ye., dorozhnyy master (stantsiya Irkutek II)

Letters to the editor. Put' i put.khos. no.11:43-45 B '58.

(NIRA 11:12)

1. Nachal'nik normativnoy stantsii tresta "Rekput'." (for Demichev).
2. Zamestitel' nachal'nika distantsii, stantsiya Kisel Sverdlovskoy
dorogi (for Koslovskiy).

(Railroad engineering)

SOV/124-58-7-8152 D

Translation from Referativnyy zhurnal, Mekhanika, 1958, Nr 7, p 119 (USSR)

AUTHOR: Kiselev, V.F.

TITLE: Methods of Strength Analysis of Spar-type and Monocoque Wings

(Metody raschetov na prochnosti lonzheronnykh i kessonnykh

kryl'yev)

ABSTRACT: Bibliographic entry on the author's dissertation for the de-

gree of Doctor of Technical Sciences, presented to the Mosk.

aviats. in-t (Moscow Aviation Institute), Moscow, 1957

ASSOCIATION: Mosk. aviats. in-t (Moscow Aviation Institute), Moscow

1. Wings--Stability 2. Wings--Analysis

Card 1/1

SOV/124-58-11-13270

Translation from: Referativnyy zhurnal, Mekhanika, 1958, Nr 11, p 197 (USSR)

AUTHOR: Kiselev, V. F. (Vladimir Falypovich)

TITLE:

Method for the Stress Analysis of a Delta Wing With Elastic Constraint (Metod rascheta treugol'nogo kryla na prochnost' s uchetom uprugoy

zadelki)

PERIODICAL: Tr. Tsentr. aero-gidrodinam. in-ta, 1957, Nr 703, 43 pp, ill.

ABSTRACT:

The stress analysis of a delta wing is reduced to the analysis of a conical shell of small elongation (aspect ratio). The solution utilizes the fundamental equations for the stresses and strains in a conical shell (ref. Balabukh, L.I., Tr. Tsentr. aerogidrodinam. in-ta, 1947, Nr 640). The influence of the constraint upon the stress distribution in the wing is accounted for with the aid of a self-balancing system of internal stresses represented in the form of a series, the terms of which consist of the products of two functions which vary along the section contour and along the generatrix of the cone. The function of the variation of the self-balancing stress system along the contour must be selected suitably, while the function of the variation of these stresses along the span must be determined from the system of Euler

Card 1/2

SOV/124-58-11-13270

Method for the Stress Analysis of a Delta Wing With Elastic Constraint

equations obtained upon variation of the expression of the potential energy of the wing. An analogous solution is obtained for a multi-web delta wing in which the webs are directed along the generatrices. In that problem the author introduces additionally the unknown stress fluxes in the spar webs which are determined from the system of canonical equations set up by the force method. Also obtained are formulas for the analysis of a swept-back box wing with the ribs aligned with the airflow; this is done by analogy with the calculation formulas obtained for the low aspect-ratio wing. The calculation procedure is exemplified in the case of the flexure of a two-web delta wing with rectangular section, loaded by afforce applied at the tip.

I. L. Kats

Card 2/2

KISELEV, V. F., Doc Tech Sci (diss) -- "Methods of computing the strength of longeron and caisson wings". Moscow, 1959. 12 pp (Min Higher Educ USSR, Moscow Order of Lenin Aviation Inst im Sergo Ordzhonikidze), 150 copies (KL, No 20, 1959, 111)

OBRAZTSOV, Ivan Filippovich; LISHLEV, VP., dotsent, kand.tekhn.nauk, retsensent; ZASLAVSAII, B.V., dotsent, kand.tekhn.nauk, red.; BOGOMCLOVA, M.F., isdat.red.; PURHLIKOVA, N.A., tekhn.red.

[Stability analysis of wing-type shell structures] Metody rascheta na prochnost' kessonnyth konstrukteii tipa kryla.

Moskva, Gos.isd-ve eber.promyahl., 1960. 311 p. (MIRA 13:5)

(Airplanes--Wings)

AM4016095

BOOK EXPLOITATION

\$12543

Kiselev, Vladimir Filippovich

Method of stress analysis of delta wing with elastic wing-root support (Metod rascheta traugol'nogo kry\*la na prochnost's uchetom uprugoy zadelki) Moscow, Oborongiz, 57. 0041 p. illus:, biblio. Errata slip inserted. No. of copies not given.

Series Note: Moscow. Tsentral'ny\*y aero-gidrodinamicheskiy institut. Trudy\*, no. 703

TOPIC TAGS: airplane wing, delta wing, stress analysis, elastic wing root support, backswept wing, multispar delta wing, shell wing, backswept ribbed wing, Castigliano method

PURPOSE AND COVERAGE: The book contains a method of determining stresses in a delta wing with elastic wing skin in the vicinity of the root support. The Castigliano variational method is used. The book is intended for engineering-technical workers in aviation design offices, and for instructors or students of higher aviation schools.

Card 7/2

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ACC NR: AP6004095 (N) SOURCE CODE: UR/0020/66/166/002/0395/0398

AUTHOR: Prudnikov, R.V.; Kiselev, V.F.; Yegorov, M.M.

ORG: Moscow State University im. M.V. Lomonosov (Moskovskiy gosudarstvennyy universitet)

TITLE: Study of the adsorptive properties of the germanium dioxide surface

SOURCE: AN SSSR. Doklady, v. 166, no. 2, 1966, 395-398

TOPIC TAGS: adsorption, germanium compound

ABSTRACT: Measurements of adsorption, heat of adsorption, structural water content, and specific surface were carried out for a germanium dioxide surface with water as the adsorbate; water was chosen because its adsorption is the most sensitive to the state of the oxide surface. The adsorptive activity of GeO<sub>2</sub> heated to various temperatures is correlated with the structural transformations taking place in this oxide; as the temperature of the heat treatment rises from 20 to 300C, the specific values of the primary adsorption increase, the maximum adsorption being displayed by samples heated to 300C: a further rise in temperature causes a sharp decrease in adsorptive activity. This behavior is attributed to the healing of surface defects and conversion to the purely tetragonal form Card 1/2

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[Strength of materials; manual for the course on "Technical mechanics, Part I" for students specializing in "Radio engineering" and "Design and construction of radio equipment"] Soprotivlenie materialov; uchebnoe posoble pokursu "Tekhnicheakaia mekhanika" chast' I dlia studentov spetsial'nostei "Radiotekhnika" i "Konstruirovanie i proizvodstvo radioelektronnoi apparatury". Izd.2., perer.

Moskva, Vses. zaochnyi energ. in-t, 1965. 389 p.

(MIRA 19:1)

MASHCHENKO, A.I.; SHIRANW, V.M.; KAZARSKIT. V.R.; KISELSV. V.).

Appearance of electron paramagnetic resonance signale during the low-temperature adsorption of various gases on reduced rutile (TiO2). Teoret. 1 eksper. khim. 1 no.31381-386 My.Je 165. (MIRA 18:9)

1. Institut khimicheskoy fiziki AN SSSR, Moskva.

KISELEV, V. G.

"Selection of Optimum Face Length Under Conditions Prevailing in the Vorkuta Coal Deposit." Cand Tech Sci, Chair for the Working of Stratified Deposits, Leningrad Order of Lenin and Order of Labor Red Banner Mining Inst, Min of Higher Education USSR, Leningrad, 1954. (KL, No 8, Feb 55)

SO: Sum. No. 631, 26 Aug 55-Survey of Scientific and Technical Dissertations Defended at USSR Higher Educational Institutions (14)

KISELEV, V.G.; KUPRIN, A.I.

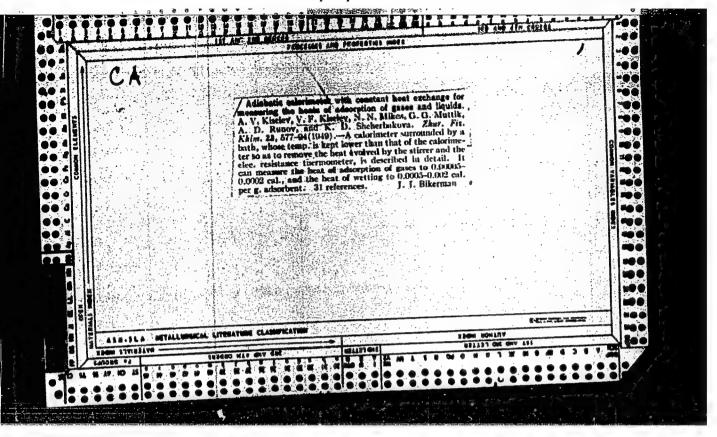
Possible flow sheets for the mining of inclined and steeply pitching seams using gravity haulage. Trudy VNIIGidrouglia no.1:64-68 '62. (MIRA 16:12)

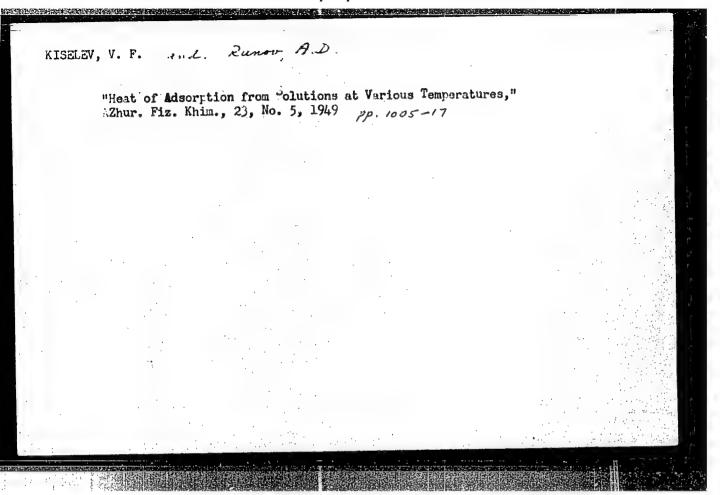
1. Sibirskiy metallurgicheskiy institut.

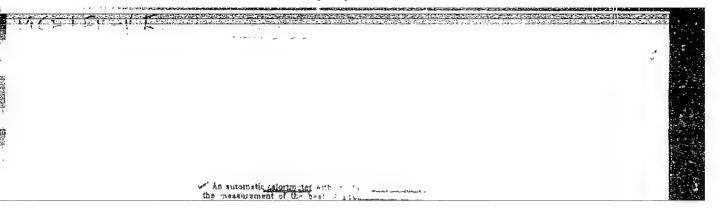
KISELEV, V.G., kand.tekhn.nauk

An important potential for increasing the efficiency of electric drills. Ugol 40 no.3:45-47 Mr 65. (MIRA 18:4)

1. Sibirskiy metallurgicheskiy institut.







KISELEV, V. F.

Cand Phys - Math See

Defended his Candidates dissertation in the Physics Faculty of Moscow State University on 2 June 1952.

Dissertation: "Heats of Adsorption by Solid Adsorbers of Pure Liquids and Solutions."

SO: Vestnik Moskovskogo Universiteta, Seriya Fiziko-Matematicheskikh i Yestestvennykh Nauk, No. 1, Moscow, Feb 1953, pp 151-157: transl. in W-29782, 12 April 54,

APPROVED FOR RELEASE: 09/17/2001

CIA-RDP86-00513R000722810011-

KISELEV, A.V.; KISELEV, V.P.; MIKOS#AVGUL, N.N.; MUTTIK G.G.; ROMOV, A.D.; SHCHERBAKOVA, K.D.;

Calorimeters and Calorimety

Automatic calorimeter with constant heat exchange for measuring heats of absorption of gases and liquids. Trudy Inst. fiz. khimii AN SSSR no. 1, 1952

Monthly List of Russian Accessions, Library of Congress, December 1952. UNCLASSIFIED

KISELEV.,\_V.\_F.

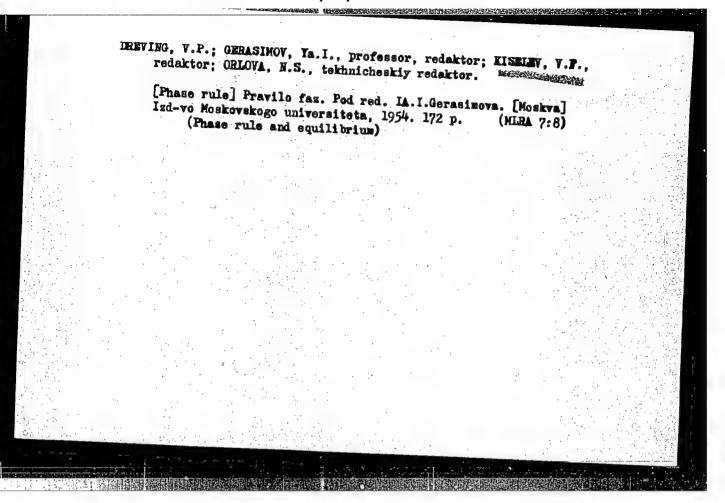
Absolute heats of wetting of strontium, lead, and barium sulfates, with water and with alcohols. B. V. Il'in and AV. F. Kiselev (Moscow State Univ.). Doklady Akad. Nauk S.S.S.R. 82, 85-7(1952 ) .-- The heats of wetting Q were detd. by calorimetry on SrSO, and PbSO, preliminarily heated to a temp. low enough  $(260^{\circ})$  to guarantee against sintering. Sp. surface areas, as detd. by the B.E.T. method of absorption of No at -195.7°, were: SrSOh 4.5, PbSOh 2.3 sq. m./g., comparing with 5.3 and 2.h by electronmicroscope photography. The mean values of 0, with  $H_2O$ , MeOH, and  $C_8H_170H$ , are, for SrSO<sub>1</sub>, 0.3h  $\pm 0.01$ , 0.22 $\pm$  0.005, and 0.23 $\pm$  0.02, and for PbSO<sub>1</sub>, 0.28 $\pm$  0.01, 0.17 $\pm$ 0.01, and 0.17  $\pm$  0.02 cal./g. The abs. values of Q, per unit surface area, are, for Srso<sub>l</sub>, 315, 200, and 215, and for Poso<sub>l</sub>, 490, 320, and 310 ergs/sq. cm. The values of Q for Srso<sub>l</sub>, Poso<sub>l</sub>, and Baso<sub>l</sub> (cf. ibid. 59, 925-7(1948); Zhur. Eksptl. Teoret. Fiz. 6, 1155(1936); C.A. 45, 3232i) are of the same order as the theoretically calcd. electro static component of the absorption energy. However, the decrease of Q from SrSO<sub>1</sub> to PbSO<sub>1</sub> to BaSO<sub>1</sub> predicted by the theory is not observed. Likewise, there is no difference in the abs. adsorption isotherms of the 3 sulfates for N<sub>2</sub> at -195.7, in miscromoles/sq. m. as a function of p/p.; the points for all 3 sulfates lie on the same isotherm. This absence of any systematic variation could be due to differences of drying conditions, or in the structure of the adsorbing surfaces; such differences appear in the electron-microscope photographs. On the other hand, there is, for each given adsorbent, a systematic decrease of Q from HoO to the alcs. The ratio of Q for H2O and for alcs. is approx. the same for SrSO, and PoSO,, 1.5-1.6. From the electrostatic theory, on the basis of the 2:1 ratio of the areas of alcs. and of HoO. a ratio of Q of 1.8-1.9 should be expected. This ratio is reduced to 1.5 if the radium of the 50, -ion is taken into account.

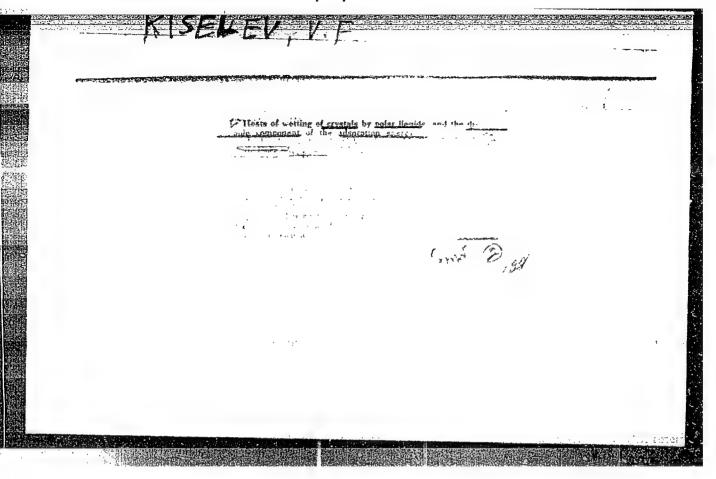
Inst. Physics, Moscow State U.

KISELEV,	V. F.	erregise (seeling			PA 234T20
		inin 5 Jul 52.	and in heat of adsorption in the region of concus- close to the point of sepn (layer formation) appar- ently are related to polymol adsorption. At that point on the adsorption isotherm, where the curve has a stepwise character, the heat of adsorption curve	"Dok Ak Nauk SSSR" Vol 86, No 1, pp 111-113 The heat of adsorption and adsorption isotherms phenol-water solns close to the sepn concn were studied. The curves for the adsorption and the of adsorption are S-shaped. Increase in adsorp	USSR/Chemistry - Adsorption  1 Sep 5%  "Studying the Adsorption and Heat of Adsorption of Phenol in Aqueous Solution on Nonporous Carbon Black V. F. Kiselev, K. G. Krasil'nikov, Moscov State Unseni M. V. Lomonosov and Inst of Phys Chem, Acad of Sci USSR
		nima. Presented	at of adsorption in the region of concust the point of sepn (layer formation) appar- related to polymol adsorption. At that the adsorption isotherm, where the curve he character, the heat of adsorption curve	5, No to to the the	Themistry - Adsorption  1 Sep 5;  ying the Adsorption and Heat of Adsorption of 1 in Aqueous Solution on Nonporous Carbon Black 1 in Aqueous Solution on Nonporous Carbon Black 1 Kiselev, K. G. Krasil'nikov, Moscov State U 1 M. V. Lomonosov and Inst of Phys Chem, Acad of SSR
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- 1. KISELEV, V. F.
- 2. USSR (600)
- 4. Carbon Black
- Absolute values of the heat of wetting of non-porous carbon black, Dokl. AN SSSR 89, no. 1, 1953.

9. Monthly List of Russian Accessions, Library of Congress, May 1953, Uncl.





KLSELEV, V.F.

USSR/ Physical Chemistry - Surface phenomena. Adsorption. Chromatography.

B-13

Abs Jour : Referat Zhur - Khimiya, No 4, 1957, 11393

Author

: Aleksandrova G.I., Kiselev V.F., Krasil'nikov K.G., Murina V.V.,

Inst Title

: Academy of Sciences USSR; Moscow State Univ. : Heat of Wetting of Silicagel of Different Degrees of Hydration by

Some Organic Liquids

Orig Pub : Dokl. AN SSSR, 1956, 108, No 2, 283-286

Abstract : Determined were the heat values of wetting of surface unit of dehydrated, at 300-900°, of specimens of silicagel (SG) of different porosity by sbsolute methanol (Q1), n-propanol (Q2) and non-polar n-heptane (Q3). Q1 does not depend on the nature of porosity of SG; Q2 and Q3 are higher in the case of coarsely porous SG, than for finely porous, which is attributed to the effect of pores which increases on transition to larger molecules of C3H7OH and C7H14. Q1 and Q2 increase linearly with degree of hy-

dration ( $^{\{i\}}_{H_2^0}$ ) of SG surface, which confirms ( see reference ) the assum-

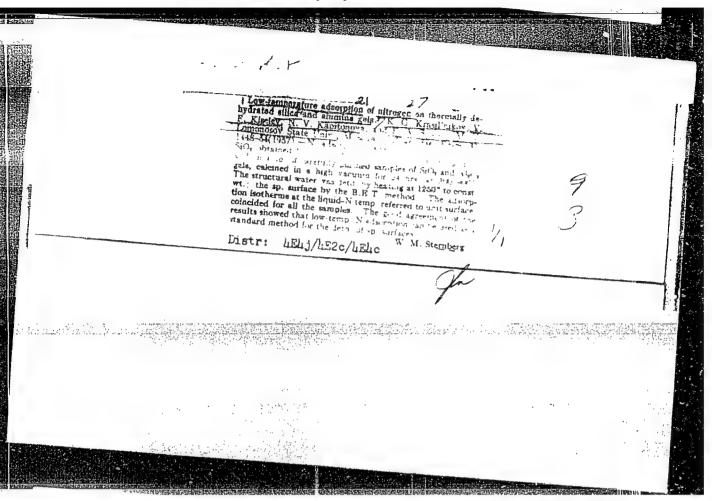
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USSR/ Physical Chemistry - Surface phenomena. Adsorption, Chromatography. Abs Jour

Referat Zhur - Philip O 17/2991 11395 CIA-RDP86-00513R000722810011-9"

ption of heterogeneity of SG surface. Q3 is almost not dependent on H20. The conclusion is arrived at that most of the earlier data on heat of wetting of SG are not mutually comparable since no account was taken of the correlation between Q and ( HgO and the nature of porosity of SG (see RZhKhim, 1956, 77773)

"APPROVED FOR RELEASE: 09/17/2001 CIA-RDP86-00513R000722810011-9



AUTHORS: Yegorov, M. M., Yegorova, T. S., Kiselev, V. F., 20-114-3-35/60 Krasil'nikov, K. G. TITLE: The Adsorption of Water Vapors on Silica Gels Hydrated to Varied Degrees (Adsorbtsiya parov vody na silikagelyakh razlichnov PERIODICAL: Doklady Akademii Nauk SSSH,1957, Vol. 114, Nr 3, pp. 579-582(USSR) ABSTRACT: As is known, the adsorption of water vapors on silica gels is characteristic by some specific properties. Some previously published scientific papers have investigated in detail the irreversible adsorption of water vapors which is connected with an additional hydration of the silica-gel surface in the process of adsorption. Other investigations reached the conclusion that the isotherm of the adsorption of water vapors, depending on the degree of the dehydration of the silica-gel surface and of porous glasses, is transformed from a convex into a concave line, the latter corresponding to a hydrophobic surface. There exist different divergences in computing the specific surfaces of silica gels from the isothermal lines. Card 1/4 None of the authors of the above-mentioned scientific papers

The Adsorption of Water Vapors on Silica Gels Hydrated to Varied Degrees

conducted the chemical analysis of the surfaces of the silica gels and of porous glasses. This task was now performed by the authors of the paper under review. Figure Nr 1 of the paper under review represents the isotherms of the water vapors on the initial silica gels and also the curves of distribution as computed from the descrption branches - of the pore volume with respect to their effective diameter taking into account the thickness of the adsorbed film. Figure Nr 2 contains the initial segments of the primary vapor adsorption on all samples of silica gels, computed for 1 m2 of the surface. It can be seen from figure Nr 2 that the isotherms of the three initial samples, worked at 300 degrees centigrade, are placed in such a way that p/p being the same, the adsorption decreases with a decrease in the degree of hydration of the surface, and this corresponding to the observed reduction in heat of the water moistening of the same samples. The state attained at the watter adsorption at the thermally dehydrated surfaces are not equilibrated, as far as in this case the process of hydration of the surface can take place. However, in the monomolecular range at small p/p this process is very slow. Therefore it is possible to consider the isotherms of the figure Nr 2A of the silica gel samples K-2; annealed at high tempera-

Card 2/4

The Adsorption of Water Vopors on Silica Gels Hydrated to Varied Degrees

tures, as equivalent from the point of view of adsorption. For this pur pose, however, one has to neglect the slight modification of the surface hydration during the process of establishing the adsorption equilibrium. If these isotherms are compared with the previous ones, it can be seen that, depending on the surface hydration, they change their form and become concave. It is furthermore observed that in this context the capacity of adsorption of the silica gel decreases. Quite a number of assumptions - as found in relevant scientific literature - on the mechanism of adsorption of water vapors on silica gel and on the hydration of its surface, are in contradiction to each other; these assumptions are based on adsorption data and also on the investigation of the infrared spectra of the surface layer. In order to clarify these questions, additional research is necessary, namely study of adsorption linked with spectroscopic investigations. There are 2 figures, 1 table, and 20 references, 14 of which are Slavic.

Card 3/4

MIDELEV, V.F.

AUTHORS:

Krasil'nikov, K. G., Kiselev, V. F., Sysoyev, Ye. A. 20-6-27/42

TITLE:

Nature of the Surface of a Dehydrated Silicagel

(K voprosu o prirode poverkhnosti degidratirovannogo

silikagelya)

PERIODICAL:

Doklady AN SSSR, 1957, Vol. 116, Nr 6, pp. 990-993 (USSR)

ABSTRACT:

The authors carried out quantitative measurements of the adsorption of nitrogen and oxygen on silicagels which were dehydrated in high vacuum. The adsorption was measured by means of the volum method. The silicagel test piece was introduced into a quartz ampule and annealed after previous draining at 300°C at an assumed temperature. Then the prepared portion of the gas to be investigated was introduced into the ampule and the corresponding measurements were carried out at 200 C. Nitrogen is not adsorbed under these conditions within the accuracy of measurement. With oxygen, the surface of silicagel dehydrated in vacuum at temperatures of 300 to 900° C adsorbes the oxygen to a considerable extent. Hereby the quantity of absorbed oxygen grows with an increase of the annealing temperature. The effect of a short-wave radiation and the thermic dehydration in the final effect apparently lead to the sameproperties of the surface. The authors further investigated the

Card 1/2

heats of wetting of the silicagels with water in which case these silicagels were previously annealed in vacuum at various temperatures up to 800° C. The data obtained during this

operation are summarized in a table. The two siling this igated here, produce after inneactor RDPR6-00513R006722610011-9'
APPROVED FOR RELEASEN 09/12/2001 nneactor RDPR6-00513R006722610011-9'
surface of the silicagel dehydrated in vacuum, centers with higher activity of adsorption than with the OF higher activity of adsorption than with the OH-groups are formed. The results obtained in this case agree with the measurements of other authors (reference 11,12).

There are 2 figures and 12 references, 8 of which are Slavic. ASSOCIATION: Moscow State University in. M. V. Lomonosov

(Moskovskiy gosudarstvennyy universitet im. M. V. Lomonosova).

May 4, 1957, by M. M. Dubinin, Academician. SUBMITTED: May 26, 1957

AVAILABLE:

Library of Congress

Card 2/2

PRESENTED:

AUTHORS: Yegorov, M.M., Yegorova, T.S., Kiselev, V.F., and Krasil'nikov, K.G. SOV/55-58-1-27/33 TITLE:

Influence of the Nature of the Silica Gel Surface on the Adsorption of the Methyl Alcohol Vapors (Vlivaniva priro PERIODICAL: V

5(4)

AUTHORS:

Il'in, B.V., Kiselev, V.F., and Krasil'nikov, K.G.

SOV/55-58-2-31/35

TITLE:

Heat of Wetting of the Silica Gels of Different Degrees of Hydration (Teploty smachivaniya silikageley razlichnoy stepeni gidratatsii)

PERIODICAL:

Vestnik Moskovskogo Universiteta, Seriya matematiki, mekhaniki, astronomii, fiziki, khimii, V./3 1958, Nr 2, pp 223-232 (USSR)

ABSTRACT:

The paper contains the results of a systematic investigation of the heat of wetting of different kinds of silica gels. The wetting of the surface was carried out by water, n-propylalcohols and n-heptane. The structural water content of the were essentially confirmed. The opinion of A.V. Kiselev and which the unit of the surface of the silica gel possesses case: The properties of the surface of the silica gel possesses case: The properties of the surface essentially depend on water content of the surface layer.

There are 6 figures, and 25 references, 15 of which are Soviet,

0000

Chair of Gen Physics Faculty of Chemistry

AUTHORS:

Kiselev, V.F., Krasil'nikov, K. G.

TITLE:

The Specific Character of the Adsorption of Phenol by Silicagel From Heptane Solutions (Osobennosti adsorbtsii fenola iz rastvorov v geptane silikagelem)

PERIODICAL:

Zhurnal fizicheskoy khimii, 1958, Vol. 32, Nr 6, pp 1435-1436

ABSTRACT:

In a previous paper it was found that the initial domain of the adsorption isothermal line shows a steplike character; more acurate measurements in this field showed that great changes of the integral heat adsorption according to the concentration take place. In connection with observations made by other authors it turned out to be interesting to carry out parallel experiments of the adsorption of phenol from solutions for purposes of investigating the adsorption isothermal line on the one hand and the heat of wetting of the same solutions on the same silicagel on the other hand. The authors used a coarse-pored silicagel KSK-1, the methods of measurement remaining the same as in the previous paper. The experimental results obtained do not yet permit the interpretation of the observations made, however, the authors put forward some ex-

The Specific Character of the Adsorption of Phenol by SOV/76-32-6-45/46

planations from which it may be seen that the phenomena are due to the complicated process of the filling of the surface of the adsorbent, which according to its properties is inhomogeneous, with the molecules of the substance to be adsorbed. It was found that the change of the chemical nature of the surface of the adsorbent caused by different ways of treatment (e.g. dehydration) can lead to the occurrence of steps in the isothermal line or to their removal, respectively. In order to be able to explain the occurrence of maxima and minima found on the isothermal line of the heat of wetting, or to find a possible connection with the step-phenomenon on the adsorption isothermal line more experiments will have to be carried out. There are 2 figures and 8 references, 7 of which

ASSOCIATION: Moskovskiy gosudarstvennyy universitet im. M.V. Lomonosova (Moscow State University imeni M.V. Lomonosov)

Card 2

5(4)

AUTHORS:

Yegorov, M. M., Krasil'nikov, K. G., Kiselev. V. F.

sov/76-32-10-33/39

TITLE:

The Influence of the Nature of Silica Gel and Quartz

Surfaces on Adsorption Properties (Vliyaniye prirody poverkhnosti silikagelya i kvartsa na ikh adsorotsionnyye svoyotva) I. Investigations of the Hydration of the Silicon Dioxide Surface (I. Issledovaniya gidratatsii poverkhnosti

PERIODICAL:

Zhurnal fizicheskoy khimii, 1958, Vol 32, Nr 10,

ABSTRACT:

Of late the presence of hydroxyl groups on silicon dioxide surfaces was found in investigations (Refs 8-13). The present paper deals in detail with investigations of the degree of hydration in dependence on the annealing in 7 different SiO<sub>2</sub> samples. The silica Gel KSK was carefully purified; silica Gel K-2 was obtained by a distillation of SiCl according to a method mentioned (Ref 3), and after storing under water it was termed silico gel K-3. "White root" and ground quartz (sample BS-1) were used as non-porous samples. The determinations

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The Influence of the Nature of Silica Cel and Quartz SCV/76-32-10-33/39 of the Silican Dioxide Surface I. Investigations of the Hydration

of the specific surfaces of the samples were carried out according to the BET method by means of nitrogen vapors. All silica gel samples used belong to the type of coarsely porous adsorbents (Ref 16). Dicgrams of the function of the water content versus the annealing temperature of the silica gels KSK-1, KSK-2, K-2 and K-3 are given using data by Shapiro and Weiss (Veys) (Ref 14) as well as by Bastick (Bastik) (Refs 4, 17). The standard temperature for treating the samples was chosen to be 300°. The results show that the content of the water of constitution as related to the surface unit is different for various silica gels. In the case where the samples were treated exactly the same but a different specific surface was present no surfaces with the same degree of hydration could be obtained, which proves the incorrectness of the data mentioned in reference 21. On storing the samples in water it was found that the amount of water of constitution on the surface increased sharply. However, those samples treated

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5(4) AUTHORS:

Yegorov, M. M., Yegorova, T. S., Krasil'nikov, K. G., 507/76-32-11-25/32

TITLE:

The Effect of the Nature of the Silica Gel and Quartz Surfacs on Its Adsorption Properties (Vliyaniye prirody poverkhnosti silikagelya i kvartsa na ikh adsorbtsionnyye svoystva) II. Adsorption of Steam, Methyl Alcohol and Nitrogen on Silica Gel of Different Degrees of Hydration (II. Adsorbtsiya parov Vody, metilovogo spirta i azota na silikagelyakh razlichnoy stepeni gidratatsii)

PERIODICAL:

Zhurnal fizicheskoy khimii, 1958, Vol 32, Nr 11, pp 2624-2633

ABSTRACT:

Silica gel samples and non-porous "white soot" described in the previous paper were used. The measurements of the adsorption were carried out according to the gravimetric method. It was found (Fig 1) that with samples treated at 300°C the adsorption (at constant p/pg) decreases with a decrease of the

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degree of hydration of the surface. The different adsorbability of the investigated silica gels is not due to their structure

The Effect of the Nature of the Silica Gel and Quartz Surface on Its on Silica Gel of Different Degrees of Hydration

but to the chemical nature of the surface (their degree of hydration). It is assumed that the hydroxyl groups with water molecules can form hydrogen compounds on the surface (Ref 12), and thus act as adsorption centers. Contradicting data given by other authors on the adsorption centers mentioned above (Refs 15, 16) are explained by a different technique of in-Vestigation. As the hydration of the surface of the investigated samples is different the adsorption properties of the surface with respect to the melecules capable of forming hydrogen compounds with hydroxyl groups are also different. Measurements carried out of the surface of hydrated KSK-1 samples occupied by water molecules showed that within the range of p/p from 0.1 to 0.3 the value  $\omega$  changes from 39 to 22.5  $\Lambda^2$  and thus is considerably higher than that given in publications (10.6 and 14.8  $\Lambda^2$ ) (Refs 20-22). As the adsorptions of the state of the s tion properties are functions of several factors (crystallography of the sample, chemical composition etc.) they cannot be called "absolute" properties ("absolute" isothermal lines). The authors thank M. M. Dubinin and B. V. Il'in.

Card 2/3

AUTHORS: Yegorov, M. M., Zarif'yunts, Yu. A., Kiselev, V. F., Krasilinikov, K. G. TI TLE: The Adsorption Properties of Alumo-Silicate Cutalynts and Their Dependence Upon Composition (Adsorbtsionny)e cveystva alyumosilikatnykh katalizatorov i ikh zavisimost! ot sostava) PERIODICAL: Doklady Akademii nauk SSSR, 1958, Vol. 120, Nr 2, ABSTRACT: In some previous papers (Refs 1-4) it was shown that the edsorption Properties per unit of surface with respect to water and ethyl alcohol molecules are to a considerable extent dependent upon the degree of hydration of the surface. It would be of interest to extend such investigations to a number of alumo-silicates of varying composition. In the first stage of these studies the authors investigated the adsorption of steam and of methyl alcohol vapors and the heat necessary to wet the synthetic alumo silicate compounds. The catalysts had a content of 15% (Gudri cutalyst ), of 30% and of 50% of Al 203. The measurements of adsorption were carried out in Card 1/3 a calorimeter with constant heat exchange. A diagram gives the

The Adsorption Properties of Alumo-Silicate Cutalyots and Their Dependence Upon Composition

SOV/20-120-2-28/63

function of the heat required for wetting by water varius the content of crystal water for all alume silicates under investigation. These curves exhibit maxima which reproduce the thermal pre-treatment of the samples at 200-300. The comparatively high content of crystal water is of interest, in particular in the samples with a high Al203 content. The heats of wetting differ by about the double between silicagel and alumo-silicate with a low Al 203 content (15%) even with a similar hydration of the surface. The same samples were also used for the dotermination of the isothermal lines of the adsorption of steam and of nethyl alcohol vapors. The description isothermal lines of all samples are considerably below the adsorption isothermal lines: Silicarel, however, did not show such a behaviour. The structure of alumosilicates is similar to that of silica, its surface, however, is more inhomogeneous. Investigation of the adsorption mechanism cannot be limited to the local adsorbed melecules with active conters, and their topography and their concentration must be taken into account. In conclusion the authors express their gratitude

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5(4) AUTHORS:

Yegorova, T. S., Kiselev, V. F., Krasil'nikov, K. G.

TITLE:

The Differential Heats of the Adsorption of Water Vapors on Silica Gels of Different Hydration (Differentsial nyye teploty adsorbtsii parov vody na silikagelyakh razlichnov gidratatsii)

PERIODICAL:

Doklady Akademii nauk SSSR, 1958, Vol 123, Nr 6, pp 1060-1063

ABSTRACT:

No reliable data have hitherto been published on the dependence of the differential adsorption heats of water vapors on the filling up of the surface. In the present paper the silica gels K - 2 and KSK - 3 were used. The characteristic data of the adsorption on these samples are given in a table. The adsorption heats of the vapors were measured in a calorimeter similar to that described by reference 7; the wetting heats were measured in a calorimeter with constant heat exchange. The water vapors were adsorbed at constant vapor pressure. The authors investigated the initial domains of isothermal lines and of the differential adsorption heats of water vapors in various silica gels by means of two methods. A diagram shows the wetting heats

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The Differential Heats of the Adsorption of Water Vapors on Silica Gels of Different Hydration

sov/20-123-5-28/50

as functions of the previously adsorbed quantity of water. In a previous paper (Ref 1) homogeneous large-pore adsorbents were investigated within the domain of adsorption up to the beginning of capillary condensation. The results obtained by calculating the differential adsorption heat as a function of specific adsorption are shown in form of a diagram. The adsorption heats for the silica gel  $K - 2 - 300^{\circ}$ , which were determined by means of direct calorimetrical measurements, agree well with the theoretically calculated curves. The initial values of water adsorption on silica gel KSK are within the interval of 15 - 20 kcal/mol. At low degrees of filling the adsorbed molecules form 3 or even 4 hydrogen bonds with the hydroxyls of the surface. Part of the molecules is probably adsorbed within this domain on centers of higher energy. In the case of one and the same degree of filling the differential heats decrease with a decreasing degree of hydration of the surface. Also the differential entropy of water vapor adsorption

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TA-PDP86-00513R00072281001

# Interaction of the surface of a solid body with water ("Interaction of the mineral part of soils with water" by

("Interaction of the mineral part of soils with water" by I.A. Tiutiunov. Reviewed by V.F. Kiselev). Pochvovedenie no.12: 105-107 D '59. (MIRA 13:4)

1. Fizicheskiy fakul'tet Moskovskogo gosudarstvennogo universiteta.

(Minerals in soils) (Soil moisture)

#### CIA-RDP86-00513R000722810011-9 "APPROVED FOR RELEASE: 09/17/2001

5(3), 5(4)

SOV/156-59-1-12/54

AUTHORS:

Kiselev, V. F., Zarif! yants, Yu. A., Kapitonova, N. V.

Krasil'nikov, K. G.

TITLE:

The Adsorption of Benzene Vapors or Aluminosilicates of Various Composition (Adsorbtsiya parov bonzola na alyumosilikatakh razlichnogo sostava)

PERIODICAL:

Nauchryye doklady vysshey shkoly. Ihimiya i khimicheskaya tekhnologiya, 1959, Nr 1, pp 48 - 1 (USSR)

ADSTRACT:

The insertion of AlO, totrahedrons in the structure of silica leads to a variation of the hydrated as well as unhydrated sectors of the surface. Thus also the adsorption properties vary during the transition from pure silica to aluminosilicates of various composition. Aluminosilicates with a content of 15, and 30% Al20, as well as

the aluminogel AT and silica gel K-2 were investigated. The isothermal lines of adsorption are given in diagrams. The initial sections (in enlarged reproduction)lie

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The Adsorption of Benzene Vapors on Aluminosilicates of Various Composition

SOV/156-59-1-12/54

on a curve, and the adsorption rises with increasing Al202 content. This cannot be explained by an increase of the adsorption potential in the pores. The adsorption of aluminogel is higher than that of equally porous aluminosilicate with 15% Al203 and of more fine-porous silica gel. The variation of the adsorptive capacity seems to depend on changes of the surface structure. This will be investigated with nonporous adsorbents in a future work. V. T. Bykov (Ref 8) assumed that the so-called "absolute" adsorption properties of the surface of silica and aluminosilicates are equal and extended this statement to various kinds of adsorbents. This is a false presumption, based on unfounded presuppositions. Actually, a function must be effective here which depends just on the specific properties of the surface of the individual adsorbents. The range, for instance, which is occupied by a benzene molecule on silica gel is larger than that on the aluminogel. Gratitude is expressed to B. V. Il'in for his assistance in this work. There are 2 figures and 16 refer-

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The Adsorption of Benzene Vapors on Aluminosilicates

sov/156-59-1-12/54

of Various Composition

ences, 14 of which are Soviet.

ASSOCIATION:

Kafedra obshchey fiziki Moskovskogo gosudarstvennogo universiteta im. M. V. Lomonosova (Chair of General Physics of

Moscow State University imeni M. V. Lomonosov)

SUBMITTED:

July 10, 1958

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CIA-RDP86-00513R000722810011-9" APPROVED FOR RELEASE: 09/17/2001

SOV/153-2-3-9/29 5(4) Yegorov, M. M., Kiselev, V. F., Krasil'nikov, K. G., AUTHORS: Simanov, Yu. P.

The Influence of the Phase Composition of the Adsorbents TITLE: in the System Al203 - H20 on Their Surface Properties

Izvestiya vysshikh uchebnykh zavedeniy. Khimiya i khimicheskaya PERIODICAL: tekhnologiya, 1959, Vol 2, Nr 3, pp 360-365 (USSR)

Cherenkov aluminum oxide from the laboratory of K. V. Topchiyeva ABSTRACT: khimicheskiy fakul tet MGU (Chemical Department of Moscow State University) was used for the investigation. The dehydration at different temperatures was investigated (Fig 1). Phase investigations were carried out by X-ray methods with cameras of the type RDK-57 and with X-ray tube of the type BSV. The samples were tempered at different temperatures and the wetting heat was determined (Table). The results are - referred to 1 g oxide - represented in diagrams (Fig 2). A second representation is given with respect to the surface unit (Fig 3). A dependence between the structural water and the wetting heat per surface unit was found (Fig 4). The phase change and the change of the degree of wetting of the surface causes a sharp change of the

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The Influence of the Phase Composition of the SOV/153-2-3-9/29 Adsorbents in the System Al<sub>2</sub>0<sub>3</sub> - H<sub>2</sub>0 on Their Surface Properties

surface properties. The authors thank K. V. Topchiyeva and B. V. Il'in for their assistance in the investigations. There are 4 figures, 1 table, and 10 references, 7 of which are Soviet.

ASSOCIATION: Moskovskiy gosudarstvennyy universitet imeni M. V. Lomonosova Kafedra fiziki (Moscow State University imeni N. V. Lomonosov

Chair of Physics)

April 24, 1958 SUBMITTED:

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CIA-RDP86-00513R000722810011-9" APPROVED FOR RELEASE: 09/17/2001

SOV/76-33-1-11/45 5(4)

Yegorov, M. M., Kiselev, V. F., Krasil'nikov, K. G., Murina, V. W. AUTHORS:

TITLE: The Effect of the Surface Nature of Silica Gel and Quartz on

Their Adsorption Properties (Vliyaniye prirody poverkhnosti silikagelya i kvartsa na ikh adsorbtsionnyye svoystva) III. Heats of Wetting of Silicon Dioxide With Various Liquids (III. Teploty smachivaniya kremnezema razlichnymi zhidkost-

PERIODICAL: Zhurnal fizicheskoy khimii, 1959, Vol 33, Nr 1, pp 65-73 (USSR)

ABSTRACT: In connection with previous papers the effect of the hydration of the surface of silicon dioxide on the adsorption energy of water and methanol in the form of heat of wetting (HW) is in-

vestigated. HW was determined in several SiO, samples with

water, methanol, n-propanol, and n-heptane in dependence on the hydration degree of the surface. Data on the HW of the silica gels KSK with water were taken from M. M. Yegorov's thesis (Ref 18). The HW was measured by means of a calorimeter with a temperature sensitivity of 5.10<sup>-50</sup>C. A table of the investigated

silica gels with the HW obtained for water is given. An in-

Card 1/3 vestigation of the effect of the glowing temperature on the HW

The Effect of the Surface Nature of Silica Gel and Quartz on Their Adsorption Properties. III. Heats of Wetting of Silicon Dioxide With Various

(Fig 1) showed that a glowing temperature of 200-300°C the function curves pass through a maximum. An increase in the glowing temperature up to 1000°C resulted in a surface decrease, e. g. in silica gel K-2, of several  $m^2/g$ . A treatment at  $300^{\circ}$ C is considered the standard. Here, the dependence of the HW on the hydration of the surface is expressed by a correction straight line. A wetting of thermally dehydrated samples with water results in the formation of hydration heat. A hydrated quartz surface differs qualitatively from a corresponding silica gel surface which can be explained by the closer packing of the hydroxyl groups (in quartz); however, investigations have still to be carried out in this respect (e. g. according to the method of the core-paramagnetic resonance). The HW of methanol does not depend on the porosity of the silica gels, which is the case with n-propanol and n-heptane. In the case of partly dehydrated surfaces a greater HW is obtained by the use of methanol than by that of water which can be explained by the effect of the methyl group in the adsorption. The results of the investigations show that the HW

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The Effect of the Surface Nature of Silica Gel and Quartz on Their Adsorption Properties. III. Heats of Wetting of Silicon Dioxide With Various

of the silica gel with water and methanol depends essentially on the hydration degree of the surface which is not the case with n-heptane. The authors thank B. V. Il'in and G. I. Aleksandrova. There are 3 figures, 1 table, and 22 references, 14 of which are Soviet.

ASSOCIATION:

Moskovskiy gosudarstvennyy universitet im. M. V. Lomonosova (Moscow State University imeni M. V. Lomonosov)

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5(4)

05806

AUTHORS:

SOV/76-33-10-4/45 Kiselev, V. F., Krasil'nikov, K. G.

TITLE:

On the Problem of the Adsorptive Power of a Unit of the

Quartz Surface

PERIODICAL:

Zhurnal fizicheskoy khimii, 1959, Vol 33, Nr 10, pp 2141-2144

ABSTRACT:

Since the quantity of adsorbed OH groups depends on the number of free corners of the SiO<sub>4</sub> tetrahedron which project into the surface of the silica-gel skeleton, it was assumed (Refs 2-4) that differences in the degree of hydration of silica gels (Refs 1-4) is connected with the menner in which the tetrahedron is packed (in dependence on the conditions of silica-gel preparation). The adsorptive properties of samples of amorphous silicon dioxide of various origin (silica gels and quartz glass) were therefore compared with those of quartz samples. since the latter has the densest packing of SiO4 tetrahedrons. The authors investigated powder samples obtained by grinding (carried out by L. A. Feygin), crystalline quartz and trans-

parent quartz glass. The samples were ground in dry state as well as under the addition of water. The adsorptive properties of the samples are listed (Table: quartz, Ky-1, -2, -3 samples, quartz glass, sample KS-1 and the silica gels KSK-1

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repruary 20, 1990

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5(4)

AUTHORS:

Bakayev, V. A., Kiselev, V. F., Krasil'nikov, K. G.

SOV/20-125-4-40/74

TITLE:

The Reduction of the Melting Temperature of Water in the Capillaries of a Porque Body (Ponizheniye temperatury plavleniya vody v kapillyarakh poristogo tela)

PERIODICAL:

Doklady Akademii nauk SSSR, 1959, Vol 125, Nr 4, pp 831-834

ABSTRACT:

From the data concerning the phase composition of an adsorbed substance as a function of temperature it is possible to determine the quantitative characteristic of the structure of a porous body by determining not only the radius but also the volume of the capillaries in which the phase transformations take place. The quantity of adsorbed substance in 1 g of the adsorbent melting at the temperature T can be determined from the specific heat of the system adsorbent—adsorbed substance. A more simple, but sensitive method is that of indirect determination of heat capacity by measuring the temperature conductivity  $\lambda$  of the system. The authors carried out these measurements by employing the modified method of "linear temperature increase". The adsorbents used were the silica gels KSK-2,

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The Reduction of the Melting Temperature of Water in the 80V/20-125-4-40/74 Capillaries of a Porous Eddy

KSM-1 and a specimen of a non-porous alumina BS-1. In these samples the isothermal lines of the adsorption of water vapors were measured. Measurements of temperature conductivity were carried out ranging from the temperature of liquid nitrogen to the temperature of 2750 K. The dependences of the quantity  $const/\lambda$  on temperature thus determined are shown by a diagram. The theory of capillary condensation shows a connection between the reduction of temperature of the phase transformation and the radius of the capillaries containing the adsorbent substance. A connection between the freezing temperature of water and the radius of the pores can be derived. The points in the diagram  $\Delta T = f(103/r)$ , which were determined for various samples and by various methods, are well suited for a straight line. The method: of determining const/\(\lambda\) suggested by the authors makes it possible quickly to determine the substance adsorbed in the porous body. Herefrom it is then possible to determine the curve for the distribution of the volume of the pores over their effective radii. The authors thank L. V. Radushkevich for his interest in this investigation.

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The Reduction of the Melting Tempeyature of Water in 807/20-125-4-40/74 the Capillaries of a Porous Body

There are 3 figures and 8 references, 2 of which are Soviet.

ASSOCIATION:

Moskovskiy gosudarstvennyy universitet im. M. V. Lomonosova (Moscow State University imeni M. V. Lomonosov). Institut fizicheskoy khimii Akademii nauk SSSR (Institute of Physical

Chemistry of the Academy of Sciences, USSR)

PRESENTED: December 24, 1958, by M. M. Dubinin, Academician

SUBMITTED: December 17, 1958

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CIA-RDP86-00513R000722810011-9" APPROVED FOR RELEASE: 09/17/2001

5 (4), 15 (2)
AUTHORS: Ganichenko, L. G., Kiselev, V. F.,
Krasil'nikov, K. G.

TITLE: The Influence of the Hydration of the Surface of Silica on the Adsorption of Aliphatic Alcohols From Solutions (Vliyaniye

gidratatsii poverkhnosti kremnezema na adsorbtsiyu

alifaticheskikh spirtov iz rastvorov)

PERIODICAL: Doklady Akademii nauk SSSR, 1959, Vol 125, Nr 6,

pp 1277-1280 (USSR)

ABSTRACT: The influence exercised by the hydration of the surface of silica is investigated for the adsorption of stream (Ref 1) and saturated hydrocarbons (Ref 2). In the former case this influence is considerable, in the latter it is insignificant. It was therefore of interest to investigate this influence in the adsorption of alcohols which have both hydroxyl groups

and carbon chains. Measurements were carried out of the adsorption of methanol-, n-propanol-, n-hexanol, and n-octanol from carbon tetrachloride solutions. Two samples of non-

porous silica - "white carbon black" - BS-1 and BS-2 were used. The samples were annealed before the experiments at 3000, one

Card 1/3 of the BS-2 samples also at 700°. The results obtained are

The Influence of the Hydration of the Surface of SOV/20-125-6-29/61 Silica on the Adsorption of Aliphatic Alcohols From Solutions

shown by table 1. Figure 1 shows the isothermal lines of adsorption, figure 2 shows the dependence a) of the adsorption maximum, b) of the surface occupied by the adsorbed molecules, c) of the thickness of the adsorption layer, d) of the ratio between the adsorbed molecules and the number of hydroxyl groups on the degree of surface hydration. Whereas methanol is still considerably influenced by the degree of hydration, this influence decreases with an increase of the carbon chain. The adsorption of octanol is not influenced at all. With an increasing length of the carbon chain the behavior of the alcohols thus approaches that of the hydrocarbons. Further, the marked increase in thickness of the adsorption layer of methanol is discussed. It is explained by variation of molecule orientation, which may be caused by a polymorphic transformation due to the thermal treatment of the silica, and leads to steps or discontinuities in the adsorption isothermal line. There are 3 figures, 1 table, and 15 references, 13 of which are Soviet.

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The Influence of the Hydration of the Surface of SOV/20-125-6-29/61 Silica on the Adsorption of Aliphatic Alcohols From Solutions

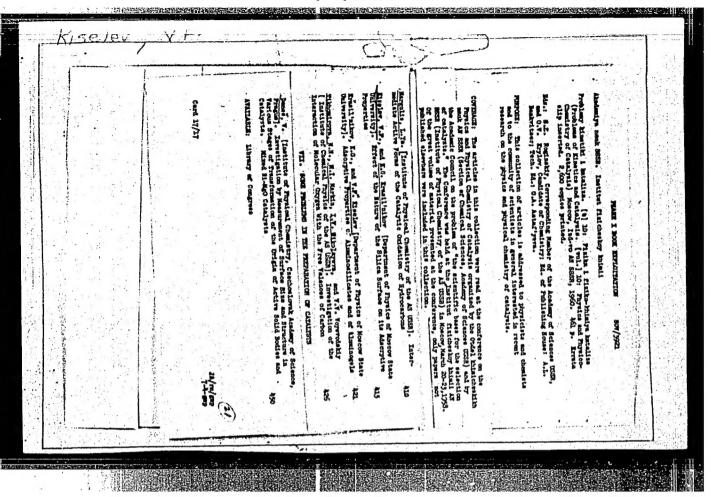
ASSOCIATION: Moskovskiy gosudarstvennyy universitet im. M. V. Lomonosova

(Moscow State University imeni M. V. Lomonosov)

PRESENTED: December 30, 1958, by M. M. Dubinin, Academician

SUBMITTED: December 24, 1958

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APPROVED FOR RELEASE: 09/17/2001 CIA-RDP86-00513R000722810011-9"

S/062/60/000/009/002/021 B023/B064

AUTHORS:

Ganichenko, L. G., Dubinin, M. M., Zaverina, Ye. D.,

Kiselev, V. F., and Krasil'nikov, K. G.

TITLE:

Study of the Vapor Adsorption on Adsorbents With

Heterogeneous Surface. Communication 2. Experiments With

Organically Substituted Silica Gel

CONTRACTOR OF THE PROPERTY OF

PERIODICAL:

Izvestiya Akademii nauk SSSR. Otdeleniye khimicheskikh

nauk, 1960, No. 9, pp. 1535-1543

TEXT: The adsorption of various vapors on methylated course-porous silica gel and a demethylated sample obtained therefrom is discussed here. The conditions of investigation were chosen in such a way that an essential change of the specific surface seemed to be unlikely. Coarse-porous commercial silica gel KCK(KSK) was taken as initial sample and carefully purified from iron and other impurities. To methylate the surface, silica gel was repeatedly treated with dichloro dimethyl silane vapors at 200°C. Then, the vapors were sucked off in vacuum at 100°C, and silica gel washed with water until the reaction for the chlorine ion was negative. The

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